A Facile Synthesis of 2-Chloro-5,7-dihydro-pyrrolo[3,4-b]pyridine-6-carboxamidine, an Annulated Nicotinoid

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The chloropyridine moiety confers high potency to several types of analgesics and insecticides acting at nicotinic acetylcholine receptors (nAChRs). This includes (−)-epibatidine 1, isolated from the skin of Ecuadorian poison frog Epipedobates tricolor.1 The pharmacological profiles of 1 show the exceptionally strong analgesic activity and the high affinity for nAChR. Recent advances in the search for nAChR ligands include the discovery of ABT-594 2, currently in development for the treatment of Parkinson’s and Alzheimer’s disease.2 Whereas, imidacloprid 3, one of the most important synthetic insecticides, acts selectively at the insect versus the mammalian nAChR.3

![Image](image-url)

The π-cation interaction plays an essential role in the central nervous system, as a number of neurotransmitters contain cationic moieties.4,5 A mounting evidence indicates the significance of the π-cation interactions in the binding of acetylcholine and related ligands to acetylcholine esterase, nicotinic and muscarinic acetylcholine receptors, and other G protein-coupled receptors. The site-directed mutagenesis analysis of nAChR provides the evidence that the highly conserved aromatic residues are possibly involved in ligand binding with a cationic center, protonated sp3 nitrogen. Indeed, the deletion or replacement of these amino acids leads to significant diminution of ligand binding affinity.

As epibatidine only contains one rotatable bond, its freedom is severely restricted. Therefore, a special attention has been paid to the preparation of conformationally constrained nicotinoids as attractive candidates for selective nAChR ligands. Thus, we designed an annulated nicotinoid, which have a chloropyridinyl nitrogen and a cationic moiety suitably placed on dihydropyrrolopyridine skeleton in order to potentially interact with the receptor sites through hydrogen bonding and electrostatic interaction, respectively. Here, we wish to report a facile synthesis of 2-chloro-5,7-dihydro-pyrrolo[3,4-b]pyridine-6-carboxamidine 12 from 2-chloroquinoline 4. In our best knowledge, a synthesis of 2-chloro-6,7-dihydro-5H-pyrrolo[3,4-b]pyridine 10 has not been reported previously.

It is envisioned that a facile synthesis of 6-chloro-pyridine-2,3-dicarboxylic acid 5 would offer the most concise synthetic route to the target molecule, as shown in Scheme 1. Despite a numerous advances in the dihydropyrrolopyridine chemistry, the synthesis of halide-containing one is quite rare, presumably, due to the lack of availability of halide-containing quinolinic and cinchomeronic acid precursors. Although there are some methods for the preparation of halide-containing pyridine dicarboxylic acids, a facile and reliable preparation of such compounds is poorly documented.5

It is well known that the ruthenium-catalyzed oxidation of aromatic rings would offer a very efficient and simple route to carboxylic acid.7 Thus, 2-chloroquinoline 4 was reacted with ruthenium tetroxide, generated in situ from RuCl3 and H2O2, in a biphasic conditions (CCl4, CH3CN, H2O). Next, the acid 5 was treated with methyl iodide and cesium carbonate in DMF to give the ester 6.8 Surprisingly, a problem was encountered in the reduction of 6. In spite of many precedences,9 an attempted reduction of 6 with BH3·THF or of 5 with LiAlH4 was unsuccessful. In both reaction conditions, the reaction mixtures turned reddish and the starting materials were completely decomposed. After several trials, we found in situ generated calcium borohydride reduction of 6 cleanly afforded the diol 7 in 88% yield.10 The remaining steps were quite straightforward. By adding of thionyl chloride, 7 was readily converted to the halide 8 in 85% yield, and subsequent substitution of 8 with p-toluenesulfonylamine and sodium hydride in DMF furnished the tosylate 9 in 57% yield. Finally, the deprotection of 9 was accomplished with hydrogen bromide (30 wt.% solution in acetic acid) in the presence of phenol to afford 2-chloro-6,7-dihydro-5H-pyrrolo[3,4-b]pyridine 10 in nearly quantitative yield.

In a recent report, Goodman et al. introduced N,N’-di-Boc-
Notes

Scheme 1. (a) RuCl₃, H₂O₂, MeCN-CCl₄-H₂O; (b) MeI, Cs₂CO₃, DMF; (c) NaBH₄, CaCl₂, THF; (d) SOCl₂; (e) H₂, NH, MeCN-CCl₄; (f) 30% HBr-AcOH, PhOH; (g) N,N′-di-Boc-N′-triflylguanidine, Et₃N, CH₂Cl₂; (h) TFA, Et₃SiH.

A round-bottom flask was charged with acetonitrile (60 mL), carbon tetrachloride (60 mL), water (90 mL), periodic acid (99.0 g, 434 mmol), and ruthenium(III) trichloride hydrate (5.0 g, 30.56 mmol) in portions, and the reaction mixture was allowed to stir for 4 h, keeping the temperature within the range of 25-40 °C by the control of the stirring speed with an ice-water bath, until no starting material was detected by TLC. The reaction mixture was cooled to 0 °C, and ether (60 mL) was added with vigorous stirring for 10 min. The organic layer was separated and the aqueous extracted with ether. The combined organic layers were washed with brine, dried over magnesium sulfate, and evaporated to yield an oil which was crystallized from dichloromethane-hexane to give 5 (4.62 g, 75%) as a solid, mp 151-152 °C (Lit. mp 152-154 °C); ¹H NMR (DMSO-d₆) δ 8.53 (d, J = 8.3 Hz, 1H), 7.70 (d, J = 8.1 Hz, 1H); EIMS m/z (rel intensity) 202 (M⁺, 2), 157 (45), 139 (100), 111 (33).

6-Chloro-pyridine-2,3-dicarboxylic acid dimethyl ester (6). Methyl iodide (24.2 g, 170.6 mmol) and cesium carbonate (27.79 g, 85.3 mmol) were added to a solution of 5 (5.73 g, 28.4 mmol) in DMF (100 mL) and the reaction mixture was allowed to stir for overnight. The mixture was diluted with ethyl acetate. The combined organic layers were washed successively with water, dried over magnesium sulfate, and evaporated under reduced pressure to give an oily residue. The residue was purified by flash chromatography on silica gel with ethyl acetate/hexane (1:7) as the eluent to give 5.28 g (81%) of 6 as a solid, mp 46-47 °C; ¹H NMR (CDCl₃) δ 7.70 (d, J = 8.1 Hz, 1H), 7.49 (d, J = 8.1 Hz, 1H), 3.96 (s, 3H), 3.93 (s, 3H); ¹³C NMR (CDCl₃) δ 157.4, 155.4, 151.3, 140.1, 125.5, 53.2; EIMS m/z (rel intensity) 231 (M⁺, 1), 198 (5), 154 (35), 102 (100), 76 (88); Anal. Calcd for C₉H₈ClNO₂: C, 47.12; H, 3.49; N, 6.02.

Experimental Section

6-Chloro-pyridine-2,3-dicarboxylic acid (5). A round-bottom flask was charged with acetonitrile (60 mL), carbon tetrachloride (60 mL), water (90 mL), periodic acid (99.0 g, 434 mmol), and ruthenium(III) trichloride hydrate (152.8 mg, 0.74 mmol). The flask contents were vigorously stirred until both phases became clear. To the flask added 2-chloroquinoline (4) (5.0 g, 30.56 mmol) in portions, and the reaction mixture was stirred for 4 h, keeping the temperature within the range of 25-40 °C by the control of the stirring speed with an ice-water bath, until no starting material was detected by TLC. The reaction mixture was cooled to 0 °C, and ether (60 mL) was added with vigorous stirring for 10 min. The organic layer was separated and the aqueous extracted with ether. The combined organic layers were washed with brine, dried over magnesium sulfate, and evaporated to yield an oil which was crystallized from dichloromethane-hexane to give 5 (4.62 g, 75%) as a solid, mp 151-152 °C (Lit. mp 152-154 °C); ¹H NMR (DMSO-d₆) δ 8.53 (d, J = 8.3 Hz, 1H), 7.70 (d, J = 8.1 Hz, 1H); EIMS m/z (rel intensity) 202 (M⁺, 2), 157 (45), 139 (100), 111 (33).

Currently, we are investigating the binding affinity of a series of compounds that contains chloropyridinyl nitrogen and a cationic moiety suitably placed on dihydropyrrolopyridine skeleton in order to potentially interact with nAChR. This work may offer a substantial method for the synthesis of halide-containing primary and secondary amines, and alcohols. Thus, the reaction of 10 with N,N′-di-Boc-N′-triflylguanidine produced N,N′-di-Boc-N′-guanidine 11, which was easily converted to the target compound 12 with trifluoroacetic acid in the presence of triethylsilane. The guanidine moiety have a strongly basic character which fully protonated under physiological conditions. Thereby, we reasoned the positive charge imposed on the molecule has a chance to form a π-cation interaction between the ligand and nAChR.

In conclusion, a highly efficient synthesis of 2-chloro-5,7-dihydro-pyrrolo[3,4-c]-pyridine-6-carboxamidine 12, an annulated nicotinoid, has been accomplished starting from 2-chloroquinoline 4. It is noteworthy that the title compound contains chloropyridinyl nitrogen and a cationic moiety suitably placed on dihydropyridopyridine skeleton in order to potentially interact with nAChR. This work may offer a substantial method for the synthesis of halide-containing 6,7-dihydro-5H-pyrrolo[3,4-b]pyridine and 2,3-dihydro-1H-pyrrolo[3,4-c]pyridine, which were difficult to obtain. Currently, we are investigating the binding affinity of a series of annulated nicotinoids to nAChR and the result will be reported in due course.
residue was purified by flash chromatography on silica gel with ethyl acetate/hexane (1 : 3) as the eluent to give 37 mg (85%) of 8. 1H NMR (CDCl3) δ 7.65 (d, J = 8.2 Hz, 1H), 7.26 (d, J = 8.2 Hz, 1H), 4.68 (s, 2H), 4.65 (s, 2H); 13C NMR (CDCl3) δ 155.4, 150.7, 141.1, 131.2, 124.8, 43.6, 41.0, EIMS m/z (rel intensity) 215 (M+, 2), 209 (M+, 49), 174 (100), 138 (42); Anal. Calcd for C18H13ClN2: C, 58.08; H, 3.90; N, 12.40; Found: C, 58.04; H, 3.87; N, 12.67.

2-Chloro-6-(toluene-4-sulfonyl)-6,7-dihydro-5H-pyrrolo-[3,4-b]pyridine (9). p-Toluensulfonamide (361 mg, 2.1 mmol) was added in portions to a suspension of 95% sodium hydride (100 mg, 4.2 mmol) in DMF (15 mL) at room temperature for 1 h and then the reaction mixture was heated to 70°C over 1 h. The mixture was diluted with ethyl acetate. The combined organic layers were washed successively with water, dried over magnesium sulfate and evaporated under reduced pressure to give an oily residue. The residue was purified by flash chromatography on silica gel with ethyl acetate/hexane (1 : 3) as the eluent to give 308 mg (57%) of 9 as a solid, mp 189-190°C; 1H NMR (CDCl3) δ 7.69 (d, J = 8.2 Hz, 2H), 7.39 (d, J = 8.0 Hz, 1H), 7.26 (d, J = 8.6 Hz, 2H), 7.12 (d, J = 8.4 Hz, 1H), 4.54 (s, 4H), 2.34 (s, 3H); 13C NMR (CDCl3) δ 157.6, 151.4, 144.1, 133.3, 130.0, 128.7, 127.6, 127.5, 123.1, 53.4, 51.7, 21.5; EIMS m/z (rel intensity) 308 (M+, 20), 153 (100), 91 (75), 65 (28).

2-Chloro-6,7-dihydro-SF-pyrrolo[3,4-b]pyridine (10). A solution of 9 (360 mg, 1.2 mmol) in 30% HBr in acetic acid (8.4 mL) and phenol (0.3 mL, 3.4 mmol) was refluxed for 0.5 h and then the solution was evaporated to dryness to yield a residual solid. The solid was collected by filtration, washed with diethyl ether, and dried under reduced pressure to give an oily residue. The residue was purified by flash chromatography on silica gel with ethyl acetate/hexane (1 : 3) as the eluent to give 264 mg (77%) of 10, mp 193°C; 1H NMR (CDCl3) δ 7.64 (d, J = 8.1 Hz, 1H), 7.54 (d, J = 8.1 Hz, 1H), 4.97 (d, J = 8.7 Hz, 4H), 1.48 (s, 18H); 13C NMR (CDCl3) δ 157.4, 154.4, 151.0, 132.7, 128.0, 126.4, 122.7, 76.9, 54.0, 52.2, 27.8; EIMS m/z (rel intensity) 396 (M+, 2), 340 (10), 284 (38), 153 (47), 57 (100).

2-Chloro-5,7-dihydro-pyrrrolo[3,4-b]pyridine-6-carboxamide (12). To a solution of 11 (226 mg, 0.57 mmol) in trifluoroacetic acid (4 mL) was added a few drops of triethylsilylane. The reaction mixture was allowed to stir for 30 min and the solution was evaporated to dryness to give a residual solid. The solid was collected by filtration, washed with diethyl ether, and dried in vacuo to yield 86.4 mg (77%) of 12, mp 240°C; 1H NMR (DMSO-d6) δ 7.89 (s, 2H), 7.81 (d, J = 6.1 Hz, 1H), 7.37 (d, J = 8.1 Hz, 1H), 7.16 (s, 1H), 4.80 (d, J = 10.4 Hz, 4H); 13C NMR (DMSO-d6) δ 156.1, 155.1, 150.4, 133.9, 127.9, 123.0, 52.4, 50.8; EIMS m/z (rel intensity) 196 (M+, 40), 153 (100), 117 (35), 69 (30).

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References