Optimizing Oily Wastewater Treatment Via Wet Peroxide Oxidation Using Response Surface Methodology

Jianzhong Shi, Xiuqing Wang, and Xiaoyin Wang†,*

College of Environmental Engineering, Wuhan Textile University, Wuhan 430073, China
†College of Science, Huazhong Agricultural University, Wuhan 430070, China.
*E-mail: wangxiuqing2000@hotmail.com

(Received December 20, 2013; Accepted December 30, 2013)

ABSTRACT. The process of petroleum involves in a large amount of oily wastewater that contains high levels of chemical oxygen demand (COD) and toxic compounds. So they must be treated before their discharge into the receptor medium. In this paper, wet peroxide oxidation (WPO) was adopted to treat the oily wastewater. Central composite design, an experimental design for response surface methodology (RSM), was used to create a set of 31 experimental runs needed for optimizing of the operating conditions. Quadratic regression models with estimated coefficients were developed to describe the COD removals. The experimental results show that WPO could effectively reduce COD by 96.8% at the optimum conditions of temperature 290°C, H₂O₂ excess (HE) 0.8, the initial concentration of oily wastewater 3855 mg/L and reaction time 9 min. RSM could be effectively adopted to optimize the operating multifactors in complex WPO process.

Key words: Wet peroxide oxidation, Oily wastewater, COD removal, Response surface methodology, Process optimization

INTRODUCTION

The oily wastewater is generated by various oil, petroleum industries, oil refineries, chemical and petrochemical plants, metal working plants, oil terminals during washing of reserving tanks. The emulsion consists of water (up to 90%), oil (mineral, organic and synthetic) and surfactant. The discharge of oily wastewater into the natural environment creates a major ecological problem throughout the world. It is necessary to purify this water so that it can be reused to save water resources and to protect the environment. It was reported that more than 2000 million tons of wastewater is produced by oil refineries only in the Middle East and European Union countries. Water quality of the oily wastewater is highly fluctuated depending on the crude oil extraction process, oil in water separation and subsequent preliminary treatment. Therefore, oily wastewater treatment is urgently needed in today’s field of environmental engineering problems. Consequently, there is a need for environmentally benign technologies capable to effectively neutralize the oily wastewater and reduce its adverse impacts on the environment.

Of the various processes that can be used to treat oily wastewater, advanced oxidation processes (AOPs) with the capability of exploiting the high reactivity of hydroxyl radicals in driving oxidation have emerged a promising technology for refractory organic compounds treatment. Several technologies such as Fenton, photo-Fenton, wet oxidation, ozonation, photocatalysis, etc. are included in the AOPs and their main differences are the source of radicals. Wet peroxide oxidation (WPO) is a kind of AOPs developed on the basis of wet oxidation and it is developed in order to decrease the running cost, which is efficient under severe temperature and pressure conditions. It has gained wide attentions of many scholars in recent years. Single-factor analysis is employed in the current study of WPO process, but it does not predict the optimum conditions well and does not also reveal the interaction between the various operating parameters. In this paper, important factors and the interaction between the factors for WPO of oily wastewater by RSM optimization are investigated. RSM is a product of the integration of mathematics and statistics, which has been widely applied to professional research in biology and statistics, but it is rarely reported in environmental governance especially in the WPO process optimization. RSM can draw the response surface according to some of point value and obtain optimal operating point through the analysis of mathematical model. But the optimal point obtained from traditional methods is only the relatively optimal point of each group in orthogonal table and is not the optimal overall value of the real.

MATERIALS AND METHODS

Experimental Apparatus and Experimental Materials

The flow diagram of the experimental setup is shown in
Optimizing Oily Wastewater Treatment Via Wet Peroxide Oxidation Using Response Surface Methodology

Fig. 1. The reactor is made from 1Cr18Ni9Ti, and its volume is 500 milliliter. It is subjected to a maximum temperature of 600 °C, a maximum pressure of 40 MPa. Oily wastewater is selected from a plant of China and the wastewater contains 37% crude oil, 61% water and 2% sediment. Oxidant is commercial H$_2$O$_2$ (analysis purity, 35 wt% aqueous solution).

Experimental Method

WPO of the oily wastewater was conducted in a laboratory-scale, batch reactor. The stirrer was used to keep from the formation of char resulting from thermal cracking of oily wastewater at the walls of the reactor.

The speed of the stirrer is 1150 r/min. The electric furnace was used as heater. The heating wires of electric furnace were placed below and around the reactor. The reaction temperature was monitored directly using thermocouple (inserted inside the reactor) and controlled within 1 °C by a temperature controller (Shanghai Yatai Instrumentation Co., N-8000 Digital Controller). Before the experiment, the reactor was loaded with oily wastewater and deionized water to bring the total volume of liquid to 90–110 mL and concentration of oily wastewater is approximately corresponding to initial COD of 1000–5000 mg/L. Then, nitrogen gas was used to purge the reactor for 10 min. After purging, the reactor was heated for about 50 min. Upon reaching the reaction condition of reaction temperatures from 250 to 330 °C, reaction temperature remained stable within 1 °C, then the specified amount of H$_2$O$_2$ was pressured and fed into the reactor using a syringe pump within 2 s at room temperature. The reaction was conducted for a given reaction time from 1 to 9 min. After the reaction, the sample valve of reactor was opened and the effluent was cooled rapidly in a shell and tube heat exchanger and then depressurized to ambient condition. The product stream was then separated into liquid and vapor phases. The liquid products were collected in a graduated cylinder. Liquid samples (ca. 15 mL) were periodically withdrawn from the reactor and analyzed.

Analytical Methods

The diluted wastewater and COD of the collected liquid are measured by potassium dichromate method of Chinese Standard 11914-89. The HE is defined as equation 1.

$$HE = \frac{(H_2O_2)_{in} - (H_2O_2)_{stoichiometric}}{(H_2O_2)_{in}} \times 100$$

Experimental Design

The central composite design, an experimental design for RSM, was used to create a set of designed experiments by MINITAB software (version 14). In this paper, the central composite design is selected for 4 factors, i.e., temperature ($X_1$), initial concentration of oily wastewater ($X_2$), reaction time ($X_3$) and HE ($X_4$).

Table 1 shows the levels of original and coded factors using central composite design. Table 2 shows central composite design and response value. Given the four main variables and five test levels, thirty-one experiments were designated by MINITAB software.

RESULTS AND DISCUSSION

RSM Model Establishment

To perform response surface regression analysis, experimental data were fitted to a full quadratic model, providing regression coefficients. The full quadratic model used in the response (COD removal) was described as the following:

$$\text{COD Removal} = 0.915743 - 0.024033X_1 + 0.030608X_2 + 0.003667X_3 + 0.009083X_4 - 0.001952X_1^2 - 0.009002X_2^2 - 0.005177X_3^2 - 0.006115X_4^2 + 0.031775X_1X_2 - 0.014212X_1X_3 - 0.012025X_1X_4 + 0.003600X_2X_3 + 0.024787X_2X_4 + 0.007775X_3X_4$$

Table 3 shows estimated regression coefficient of COD
removal and Table 4 shows the results of variance analysis for COD removal. From Table 4, it is seen that the influence of each independent factor on the models was tested for its level of statistical significance by analyzing the variance. These results show that the effects of all terms on COD removal are statistically significant at the 5% level. So it is concluded that the affecting factors for COD removal are more complex and it is joint action by many factors.

**Interaction by RSM**

The surface plot of objective function is drawn by Minitab in order to more directly reflect the interaction between the various factors that affect the COD removal and the role of strength for interaction between the various experimental factors.

**Table 2. Central composite design and response value**

<table>
<thead>
<tr>
<th>Number</th>
<th>X₁</th>
<th>X₂</th>
<th>X₃</th>
<th>X₄</th>
<th>COD Removal / %</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>1</td>
<td>1</td>
<td>−1</td>
<td>−1</td>
<td>90.59</td>
</tr>
<tr>
<td>2</td>
<td>−2</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>86.67</td>
</tr>
<tr>
<td>3</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>95.33</td>
</tr>
<tr>
<td>4</td>
<td>1</td>
<td>−1</td>
<td>1</td>
<td>−1</td>
<td>80.00</td>
</tr>
<tr>
<td>5</td>
<td>−1</td>
<td>1</td>
<td>−1</td>
<td>1</td>
<td>94.00</td>
</tr>
<tr>
<td>6</td>
<td>0</td>
<td>0</td>
<td>−2</td>
<td>0</td>
<td>88.00</td>
</tr>
<tr>
<td>7</td>
<td>−1</td>
<td>−1</td>
<td>−1</td>
<td>−1</td>
<td>89.75</td>
</tr>
<tr>
<td>8</td>
<td>0</td>
<td>1</td>
<td>−1</td>
<td>−1</td>
<td>91.44</td>
</tr>
<tr>
<td>9</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.15</td>
</tr>
<tr>
<td>10</td>
<td>0</td>
<td>0</td>
<td>−2</td>
<td>0</td>
<td>86.00</td>
</tr>
<tr>
<td>11</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.85</td>
</tr>
<tr>
<td>12</td>
<td>0</td>
<td>0</td>
<td>−2</td>
<td>0</td>
<td>88.00</td>
</tr>
<tr>
<td>13</td>
<td>0</td>
<td>0</td>
<td>−2</td>
<td>−2</td>
<td>80.00</td>
</tr>
<tr>
<td>14</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>15</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>−2</td>
<td>84.00</td>
</tr>
<tr>
<td>16</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.85</td>
</tr>
<tr>
<td>17</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>18</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>19</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>20</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>21</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>22</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>23</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>24</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>25</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>26</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>27</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>28</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>29</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
<tr>
<td>30</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>91.44</td>
</tr>
</tbody>
</table>

**Table 3. Estimated regression coefficient of COD removal**

<table>
<thead>
<tr>
<th>Designate Coefficients</th>
<th>Standard error of coefficients</th>
<th>t-value</th>
<th>p</th>
</tr>
</thead>
<tbody>
<tr>
<td>Constant</td>
<td>0.915743</td>
<td>0.011783</td>
<td>77.717</td>
</tr>
<tr>
<td>X₁</td>
<td>−0.024033</td>
<td>0.006364</td>
<td>−3.777</td>
</tr>
<tr>
<td>X₂</td>
<td>0.030608</td>
<td>0.006364</td>
<td>4.810</td>
</tr>
<tr>
<td>X₃</td>
<td>0.003667</td>
<td>0.006364</td>
<td>0.576</td>
</tr>
<tr>
<td>X₄</td>
<td>0.009083</td>
<td>0.006364</td>
<td>1.427</td>
</tr>
<tr>
<td>X₁X₂</td>
<td>−0.001952</td>
<td>0.005830</td>
<td>−0.335</td>
</tr>
<tr>
<td>X₁X₃</td>
<td>−0.009002</td>
<td>0.005830</td>
<td>−1.544</td>
</tr>
<tr>
<td>X₁X₄</td>
<td>−0.005177</td>
<td>0.005830</td>
<td>−0.888</td>
</tr>
<tr>
<td>X₂X₃</td>
<td>−0.006115</td>
<td>0.005830</td>
<td>−1.049</td>
</tr>
<tr>
<td>X₂X₄</td>
<td>0.031775</td>
<td>0.007794</td>
<td>4.077</td>
</tr>
<tr>
<td>X₃X₄</td>
<td>−0.014212</td>
<td>0.007794</td>
<td>−1.824</td>
</tr>
<tr>
<td>X₁X₂X₃</td>
<td>−0.012025</td>
<td>0.007794</td>
<td>−1.543</td>
</tr>
<tr>
<td>X₁X₂X₄</td>
<td>0.003600</td>
<td>0.007794</td>
<td>0.462</td>
</tr>
<tr>
<td>X₁X₃X₄</td>
<td>0.024787</td>
<td>0.007794</td>
<td>3.180</td>
</tr>
<tr>
<td>X₃X₄</td>
<td>0.007775</td>
<td>0.007794</td>
<td>0.998</td>
</tr>
</tbody>
</table>

**Table 4. The results of variance analysis for COD removal**

<table>
<thead>
<tr>
<th>Source</th>
<th>Degree of freedom</th>
<th>Sum of squares</th>
<th>Adjusted mean squares</th>
<th>F-value</th>
<th>p</th>
</tr>
</thead>
<tbody>
<tr>
<td>Regression</td>
<td>14</td>
<td>0.074848</td>
<td>0.005346</td>
<td>5.50</td>
<td>0.001</td>
</tr>
<tr>
<td>Interaction</td>
<td>6</td>
<td>0.032705</td>
<td>0.005451</td>
<td>5.61</td>
<td>0.003</td>
</tr>
<tr>
<td>Residual error</td>
<td>16</td>
<td>0.015550</td>
<td>0.000972</td>
<td>132.31</td>
<td>0.000</td>
</tr>
<tr>
<td>Lack-of-fit</td>
<td>10</td>
<td>0.015480</td>
<td>0.001548</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pure error</td>
<td>6</td>
<td>0.000070</td>
<td>0.000012</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total</td>
<td>30</td>
<td>0.090398</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**Figure 2.** Response surface plots of the interaction between temperature and initial concentration of oily wastewater on COD removal.
Optimizing Oily Wastewater Treatment Via Wet Peroxide Oxidation Using Response Surface Methodology

...ture, initial concentration of oily wastewater, reaction time, HE. In each plot, two factors are varied, while the rest is kept constant.

For Fig. 2, it is shown that temperature and initial concentration have obvious effects on COD removal. When temperature is closed to 330 °C and initial concentration of oily wastewater is about 1000 mg/L, COD removal is about 70%. However, temperature is closed to 250 °C and initial concentration of oily wastewater is about 5000 mg/L, COD removal is above 90%. For Fig. 3, it is shown that higher HE is, more obvious COD removal is. When HE is closed to 0.8, COD removal is about 93%. Effect of reaction time on COD removal changes little. For Fig. 4, it is seen that there is strong interaction between initial concentration of oily wastewater and HE. When HE increases, COD removal increases. However, COD removal reduces when initial concentration of oily wastewater increases.

For Fig. 5, it is shown that COD removal increases when initial concentration of oily wastewater and reaction time both increase. Increased range of reaction time on COD removal is not very obvious. For Fig. 6, it is seen that there is strong interaction between temperature and HE. HE plays the main role in COD removal when temperature changes little. For Fig. 7, it is shown that there is interaction between temperature and reaction time. Increased range of reaction time on COD removal is not very obvious.

Optimization of Process Parameters Using RSM

Objective function of Minitab’s response surface provides an intuitive tool for objective optimization and its unique response optimizer is a powerful tool for multi-objective problem to the solution encountered in the experimental design. And the target function is optimized by Minitab’s response optimizer. The result is shown in Fig. 8. It is calculated that the optimized conditions are that temperature is 290 °C, HE is 0.8, the initial concen-
Concentration of oily wastewater is 3855 mg/L and reaction time is 9 min.

CONCLUSIONS

This study demonstrates that the WPO process is effective in reducing COD by 96.8%. Therefore, the WPO process is possibly suitable for a primary treatment for oily wastewater. Central composite design provides sufficient data to fit the quadratic models for COD removals. The optimization of the models provides the optimum conditions: temperature is 290 °C, HE is 0.8, the initial concentration of oily wastewater is 3855 mg/L and reaction time is 9 min. RSM could be effectively adopted to optimize the operating multifactor in complex WPO process.

Acknowledgments. This work was supported by Hubei Provincial Department of Education “Evaluation of textile fiber dust explosion performance and explosion danger grade” (D20121706), (T201207). Young and Mid-aged Scientific and Technical Innovation Team of Colleges and Universities in Hubei (ID: T201207). And the publication cost of this paper was supported by the Korean Chemical Society.

REFERENCES