Influence of MWCNTs on Fracture Toughness of MWCNTs/Nickel-Pitch Fiber/Epoxy Composites

Yoon-Ji Yim*, Soo-Jin Park†

ABSTRACT: The influence of MWCNTs on fracture toughness properties of MWCNTs/Nickel-Pitch Fibers/epoxy composites (MWCNTs/Ni-PFs/epoxy) was investigated according to MWCNTs content. Nickel-Pitch-based carbon fibers (Ni-PFs) were prepared by electroless nickel-plating. The surface properties of Ni-PFs were determined by scanning electron microscopy (SEM) and X-ray photoelectron spectrometry (XPS). The fracture toughness of MWCNTs/Ni-PFs/epoxy was assessed by critical stress intensity factor (\(K_{IC}\)) and critical strain energy release rate (\(G_{IC}\)). From the results, it was found that the fracture toughness properties of MWCNTs/Ni-PFs/epoxy were enhanced with increasing MWCNTs content, whereas the value decreased above 5 wt. % MWCNTs content. This was probably considered that the MWCNTs entangled with each other in epoxy due to an excess of MWCNTs.

Key Words: Pitch Fibers, Multi-walled Carbon Nanotubes (MWCNTs), Electroless Nickel-plating, Epoxy Composites, Fracture Toughness

1. INTRODUCTION

These days, pitch-based carbon fibers (pitch fibers) have received considerable attention due to their advantages such as high carbon yield, low cost, and good graphitizability that are characteristic of pitches derived from coal tar or petroleum. These excellent properties of pitch fibers are transferred to matrix through their interface. Accordingly most of pitch-based carbon fibers reinforced composite properties are controlled by interfacial adhesion between pitch fibers and matrix resins [1,2].

Carbon nanotubes (CNTs) are a typically nanofillers and were discovered as fullerene-related structures by Iijima at NEC laboratory of Tsukuba in 1991 [3]. Carbon nanotubes also have several excellent properties including electronic conductivity, thermal conductivity, corrosion resistance, high strength, Young's modulus and so on [4-6].

In this study, epoxy matrix was reinforced with Pitch fibers and multi walled carbon nanotubes (MWCNTs) to enhance the fracture toughness of the composites. In order to improve interfacial adhesive strength between fillers and matrix resins, the filler surface is commonly modified by numerous surface treatments [7-9]. Therefore the pitch fibers and MWCNTs were electrolessly nickel-plated and acidic treated, respectively.

The objective of this study is to evaluate the influence of MWCNTs on the fracture toughness of MWCNTs/Ni-PFs/epoxy composites. The pitch fiber surfaces were characterized by SEM and XPS to confirm the morphology of the fiber surfaces and the elemental compositional changes. The critical stress intensity factor (\(K_{IC}\)) and critical strain energy release rate (\(G_{IC}\)) were investigated to determine the fracture toughness of the MWCNTs/Ni-PFs/epoxy composites according to MWCNTs contents.

2. EXPERIMENTAL

2.1 Specimen preparation

The epoxy resin used in this work was diglycidyl ether of bisphenol A (DGEBA, YD-128) which was supplied by the Kukdo Chem. Co., Korea. The curing agent used in this work was a modified cycloaliphatic amine curing agent (KH-819) supplied by the Kukdo Chem. Co., Korea. Pitch fibers (PFs)
were supplied by GS Caltex Co., Korea. MWCNTs were used as a reinforcement material and were manufactured by the CVD process (purity: 95 wt.%, diameter: ~10 nm, length: 10–20 μm) which was supplied by Nanosolution, Korea.

The schematic diagram of the electroless nickel-plating processes involved in the activation and metal deposition is illustrated in Fig. 2(a). Before the electroless nickel-plating process, the PFs were pretreated in 5M HNO₃ for 30 min in order to increase interfacial adhesion between the nickel and the PFs. They were sequentially activated in tin chloride (SnCl₂) and palladium chloride (PdCl₂) solution for 30 min each. By activating, Sn/Pd nucleates adhered to the surface of PFs, and these nucleates accelerated the metal plating during the electroless nickel-plating. Ni-PFs were obtained by dipping the PFs into the nickel bath for 300 s. The constituents of plating solutions and reaction conditions are listed in Table 1.

The MWCNTs were acid treated by following method before use. The acid treatments were based on a mixture of concentrated nitric (HNO₃) and sulphuric acids (H₂SO₄) in a ratio of 1:3, respectively. In a typical experiment, a 1 g portion of as-received MWCNTs was added to 40 cm³ of the acid mixture in a round-bottomed flask, and refluxed for 30 min. On cooling, the mixture was washed with distilled water on a sintered glass filter until the washings showed no acidity.

The Ni-PFs (5 wt.%) and MWCNTs (1, 2, and 5 wt.%) were dispersed within the epoxy resin for 2 h using a 3-roll-mill at room temperature. The 3-roll-mill is a machine that uses shear force created by three horizontally positioned rolls rotating in opposite directions and different speeds relative to each other, in order to mix, refine, disperse, or homogenize viscous materials fed into it (Fig. 1). The mixtures were then poured into a mold and cured at 100°C for 1 h and 130°C for 1 h.

### Table 1. Composition and Conditions of Electroless Ni Plating Bath

<table>
<thead>
<tr>
<th>Composition &amp; Condition</th>
<th></th>
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<tbody>
<tr>
<td>NiSO₄·6H₂O</td>
<td>280 g/L</td>
</tr>
<tr>
<td>NiCl₂·6H₂O</td>
<td>40 g/L</td>
</tr>
<tr>
<td>Na₃C₆H₅O₇·1.5H₂O</td>
<td>15 g/L</td>
</tr>
<tr>
<td>NaH₂PO₂·2H₂O</td>
<td>100 g/L</td>
</tr>
<tr>
<td>NH₄Cl</td>
<td>100 g/L</td>
</tr>
<tr>
<td>PbNO₃</td>
<td>30 g/L</td>
</tr>
<tr>
<td>pH</td>
<td>8.25</td>
</tr>
<tr>
<td>Temperature (°C)</td>
<td>90 ± 1</td>
</tr>
<tr>
<td>Plating time (sec)</td>
<td>300</td>
</tr>
</tbody>
</table>

2.2 Characterization

The morphologies of the Ni-PFs were measured by a scanning electron microscope (SEM, JSM-6701F, JEOL). Fourier transform infrared (FT-IR) spectra of as-received MWCNTs and acidic MWCNTs (A-MWCNTs) were recorded with a FT-IR spectrometer (Nicoleti S10, Thermo SCIENTIFIC, USA). The surface properties of the Ni–PFs were characterized by X-ray photoelectron spectroscopy (XPS, K-alpha, Thermo Scientific). The XPS experiment was performed using a Kα spectrometer equipped with an AlKα X-ray source. The base pressure in the sample chamber was controlled in the range from 10⁻⁸ to 10⁻⁹ torr. The critical stress intensity factor (Kᵥc) and critical strain energy release rate (Gᵥc) were measured using a single-edge-notched (SEN) beam fracture toughness test in a three-point flexural test. The fracture toughness tests were conducted using a universal test machine (UTM, LR5K plus, Lloyd) in accordance with the ASTM D-5045-95. The span-to-depth ratio and crosshead speed were 4:1 and 1 mm/min. The specimen dimensions were 50 mm × 10 mm × 5 mm. The notches were cut using a diamond coating saw, approximately half the depth of the specimen. The fracture toughness values reported represent an average of the results for tests run on at least six specimens.

3. RESULTS AND DISCUSSION

The SEM images of as-received and Ni-PFs are shown in Fig. 2(b). The surface morphologies of as-received PFs are smooth and clean. On the other hand, the surfaces morphologies of Ni-PFs shows that nickel particles were homogeneously plated on the surfaces of Ni-PFs [10].

The FT-IR spectra and SEM image of MWCNTs are shown in Fig. 3. The FT-IR spectra of MWCNTs showed peaks with very low intensity at 3432 cm⁻¹ corresponding to O-H group. In the case of A-MWCNTs, this band appeared with significantly higher intensity according to the degree of modification. This was attributed to the increased number of carboxylic acid groups generated at the surface of the MWCNTs after acidic treatments [11]. The acidic treatment strengthens the peak intensity because of increase in active groups on the MWCNTs. These active groups may help to change the polar-
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It is well known that XPS is a powerful analytical technique to confirm the chemical composition of materials. The XPS spectra and elemental compositions of as-received PFs and Ni-PFs are presented in Fig. 4 and Table 2, respectively. As a result, the spectrum of as-received PFs shows an intense C1s peak and a O 1s peak at 284.6 and 532.8 eV, respectively. The O1s peak is probably due to intrinsic surface carboxyl groups.

By contrast, C1s, O 1s, and Ni 2p (B.E. = 857.6 eV) peaks are observed in XPS spectra of Ni-PFs. The O1s peak of Ni-PFs is probably because of PFs surfaces of NiO, C=O, O-C-O, –OH, and -C-O groups [12]. From the XPS results, it is found that the surface composition of the PFs changed considerably after the electroless nickel plating. The carbon content of Ni-PFs decreased, whereas the oxygen and nickel contents of Ni-PFs increased compared with those of PFs. The O1s/C1s ratios of Ni-PFs are also increased compared to those of as-received PFs [13]. It is thought that the Ni-plating formed deposition of more active groups on the inactive carbon, thereby enhancing the surface functionality of Ni-PFs.

The understanding of fracture phenomenon was first presented by Griffith [14]. It is generally accepted that the fracture toughness of filler-reinforced composites depend strongly on the level of adhesion between the filler and the matrix. The fracture toughness of MWCNTs/Ni-PFs/epoxy as a function of MWCNTs content was determined in terms of the critical stress intensity factor (KIC). The values of KIC were calculated by the following equations [15,16].

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**Fig. 2.** (a) Schematic diagrams of the electroless Ni-plating processes involved in the activation and metal deposition; (b) SEM images of as-received PFs and 300 Ni-PFs

**Fig. 3.** (a) FT-IR spectra of as-received MWCNTs and A-MWCNTs; (b) SEM image of MWCNTs

**Fig. 4.** XPS spectra of (a) as-received PFs and (b) Ni-PFs

**Table 2.** Elemental compositions of as-received PFs and Ni-PFs

<table>
<thead>
<tr>
<th>Samples</th>
<th>C1s (%)</th>
<th>O1 (%)</th>
<th>Ni2p (%)</th>
<th>O1s/C1s</th>
</tr>
</thead>
<tbody>
<tr>
<td>as-received PFs</td>
<td>80.1</td>
<td>16.5</td>
<td>-</td>
<td>0.2</td>
</tr>
<tr>
<td>Ni-PFs</td>
<td>41.0</td>
<td>44.4</td>
<td>12.6</td>
<td>1.0</td>
</tr>
</tbody>
</table>
crack length, width and thickness, respectively. where and

\[ K_{IC} = \frac{P \cdot L}{b \cdot d^{3/2}} \cdot Y \]  

(1)

and

\[ Y = \frac{3(a/d)^{2/3}[1.99-(a/d)(1-a/d)(2.15-3.93a/d+2.7a^2/d^2)]}{2(1+2a/d)(1-a/d)^{2/3}} \]  

(2)

where \( P \) is the rupture force, \( L \) the span between the supports, and \( Y \) the geometrical factor; \( a, b \) and \( d \) are the specimen pre-crack length, width and thickness, respectively.

The values of \( G_{IC} \) can be calculated from the results of the \( K_{IC} \). Using the Irwin relationship [17], the equation (3) according to stress and strain under the same conditions was used.

\[ G_{IC} = \frac{K_{IC}^2}{E} \cdot (1 - \nu^2) \]  

(3)

where \( \nu \) is the Poisson’s ratio (\( \nu \approx 0.3 \)), and \( E \) is the Young’s modulus (flexural modulus) obtained from flexural testing. The Young’s modulus is the ratio of stress to strain in flexural deformation.

The influence of MWCNTs on fracture toughness of MWCNTs/Ni-PFs/epoxy can be seen in Fig. 5. Fig. 5(a) shows the values of \( K_{IC} \) for MWCNTs/Ni-PFs/epoxy as a function of MWCNTs content. Firstly, \( K_{IC} \) values of Ni-PFs/epoxy composites are higher than as-received PFs/epoxy composites. Meanwhile, it is observed that fracture toughness were decreased above certain content. It seems that MWCNTs entangled with each other in the matrix due to an excess of MWCNTs.

4. CONCLUSIONS

The effects of the concentration of MWCNTs (2~5 wt.%) on fracture toughness of MWCNTs/Ni-PFs/epoxy composites were investigated. Our experimental results suggested that mechanical properties were gradually increased according to increasing MWCNTs content. The addition of MWCNTs plays a significant role in the reinforcement of the Ni-PFs/epoxy composites. Meanwhile, it is observed that fracture toughness were decreased above certain content. It seems that MWCNTs entangled with each other in the matrix due to an excess of MWCNTs.

ACKNOWLEDGEMENT

This research was financially supported by the “Carbon valley construction program” through the Ministry of Trade, Industry & Energy (MOTIE) and Korea Institute for Advancement of Technology (KIAT).

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