Purification and Characterization of β-Xylosidase from *Paenibacillus* sp. DG-22

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An intracellular β-xylosidase from *Paenibacillus* sp. DG-22 was purified to homogeneity by ion-exchange, hydrophobic interaction and gel-filtration chromatography. The molecular weight of the enzyme was measured to be 156,000 by gel filtration and 80,000 by SDS-PAGE, indicating that the enzyme consisted of two identical subunits. The purified enzyme exhibited maximum activity at 65°C and pH 5.5. It retained 80% of its initial activity up to 60 min at 60°C and had a half-life of 25 min at 65°C. The enzyme was highly specific for pNPX as the substrate. It showed little or no activity against other p-nitrophenyl glycosides and xylans. The *Km* and *Vmax* for pNPX was 0.53 mM and 3.18 U/mg protein, respectively. The β-xylosidase was strongly inhibited by Ag⁺, Fe³⁺, Hg²⁺ and Zn²⁺ and slightly activated by DTT. The hydrolysis product from xylobiose, xylotriose, and xylotetraose was xylose.

**Key words** – *Paenibacillus* sp., β-xylosidase, purification

**Introduction**

Xylan is the major component of plant hemicellulose and the second most abundant renewable polysaccharide in nature. It is a heterogeneous polysaccharide consisting of a main chain of β-1,4-linked D-xylose residues with arabinofuranose, glucuronic acid and acetyl side groups. The composition and structure of xylan vary according to the source [4]. Complete breakdown of xylan requires the cooperative action of several hydrolytic enzymes of which endoxylanase and β-xylosidase are the most important [2]. Endoxylanase (1,4-β-D-xylan xylohydrolase, EC 3.2.1.8) randomly cleaves off the β-1,4 bonds in the xylan backbone to yield xylo-oligosaccharides. β-Xylosidase (1,4-β-D-xylan xylohydrolase, EC 3.2.1.37) hydrolyzes xylobiose and short chain xylo-oligosaccharides from the non-reducing end to produce xylose, which can then be used as a fermentation feed stock for various useful products such as xylitol and ethanol [22,23]. It also plays an important role in relieving the end product inhibition of endoxylanase, therefore it is essential for complete breakdown of xylan [17]. Accessory enzymes, such as α-arabinofuranosidase, α-glucuronidase and acetylxylan esterase are also required for the removal of side groups [4].

Although many fungal β-xylosidases have been studied [8,18], only a few reports are available on the purification and characterization of bacterial β-xylosidases [10,12,19], including those expressed in *E. coli* [7] and *Saccharomyces cerevisiae* [5]. The β-xylosidases are intracellular in most bacteria and yeasts, whereas those of fungi are secreted into culture media [18]. β-Xylosidase have been classified into families 39, 43 and 52 of glycosyl hydrolases, based on their amino acid sequence similarities [6].

*Paenibacillus* sp. DG-22, a moderately thermophilic bacterium isolated from timber yard soil, grows actively on xylan as a sole carbon source and does not have cellulase activity [11]. We previously reported the purification and characterization of endoxylanases [15] and cloning of xylanase A gene from this bacterium [13]. The synthesis of β-xylosidase from *Paenibacillus* sp. DG-22 was induced by xylan and methyl β-D-xylopyranoside but repressed by glucose and xylose [14]. The objective of this work is to purify and characterize the intracellular β-xylosidase from *Paenibacillus* sp. DG-22.

**Materials and Methods**

**Chemicals**

Beechwood xylan, birchwood xylan, oat spelts xylan, methyl β-D-xylopyranoside, p-nitrophenyl glycosides and xylose were purchased from Sigma Chemical Co. (St. Louis, MO, USA). Xylobiose (X₂), xylotriose (X₃) and xylotetraose (X₄) were obtained from MegaZyme (North Rocks, Australia). Q-Sepharose, Sephacryl S-200 and Sephacryl S-300 were purchased from Amersham Pharmacia Biotech (Piscataway, NJ, USA). Butyl Spheroilose was obtained from
Enzyme and protein assay

β-Xylosidase activity was measured by a spectrophotometric method with 4-nitrophenyl-β-D-xylopyranoside (pNPP) as the substrate. The reaction mixture, composed of 1 mM pNPP, 50 mM sodium acetate (pH 5.5), and diluted enzyme in 0.5 ml reaction volume, was incubated at 60°C for 10 min. The reaction was stopped by the addition of 1.0 ml of 1.0 M sodium carbonate and the nitrophenol released was measured as absorbance at 410 nm. A standard curve was prepared by using p-nitrophenol. One unit of β-xylosidase activity was defined as the amount of enzyme which released 1 μmol of p-nitrophenol in 1 min. Other p-nitrophenol derivatives of sugars were also tested as substrates under the same conditions.

Protein concentration was determined by the Bradford method [3] with the Bio-Rad protein assay reagent (Bio-Rad Laboratories) using bovine serum albumin as the standard.

Enzyme purification

Unless otherwise stated, all steps were conducted at room temperature. Cells grown on one liter of 2x YT medium supplemented with 1% (w/v) methyl β-D-xylopyranoside were suspended in 30 ml of 20 mM sodium phosphate (pH 7.0), and sonicated on ice for 20 min with a Branson sonifier (250 watts, 40% pulsed mode). The supernatant obtained by centrifugation (12,000x g, 20 min) was used as the crude enzyme preparation.

(i) Ion-exchange chromatography: The crude enzyme solution was put on a Q-Sepharose column (2.5 × 18 cm) pre-equilibrated with 20 mM sodium phosphate buffer (pH 7.0). The column was washed with the same buffer using two-folds volume of bed volume and absorbed proteins were eluted with a linear gradient of 0 to 0.8 M NaCl in the same buffer at a flow rate of 1.0 ml/min. The active β-xylosidase peak fractions were concentrated by ultrafiltration using a YM30 membrane (Amicon, Beverly, USA) and dialyzed against 20 mM sodium phosphate buffer (pH 7.0) containing 0.6 M ammonium sulfate.

(ii) Hydrophobic interaction chromatography: The dialyzed enzyme solution was then applied to a Butyl Spherose column (2.5 × 5 cm), previously equilibrated with 20 mM sodium phosphate buffer (pH 7.0) containing 0.6 M ammonium sulfate. The column was washed with the same buffer and eluted with decreasing ammonium sulfate gradient (0.6-0 M) at a flow rate of 0.8 ml/min. The active fractions were dialyzed overnight against 50 mM sodium acetate buffer (pH 5.5) containing 0.2 M NaCl and concentrated by ultrafiltration.

(iii) Gel-filtration chromatography: The concentrated enzyme solution was applied to a column of Sephacryl S-300 (2.5 × 50 cm) pre-equilibrated with 50 mM sodium acetate buffer (pH 5.5) containing 0.2 M NaCl. The enzyme was eluted with the same buffer at a flow rate of 0.24 ml/min. Active fractions were pooled and concentrated by ultrafiltration.

Estimation of molecular weight

The apparent molecular weight of the native enzyme was determined by gel filtration on a Sephacryl S-200 column (1.0 × 50 cm). The molecular weight determination under denaturing condition was performed by sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) according to Laemmli [9]. Gel (12.5%) was stained with Coomassie Brilliant Blue R-250.

Effect of pH and temperature

The effect of pH on the enzyme activity was estimated using assay in McIlvaine buffer (0.2 M Na2HPO4-0.1 M citric acid) at pH values from pH 3.0 to 7.5 at 60°C. The effect of temperature on the activity was assayed by incubating 0.1 U of the purified enzyme in sodium acetate buffer (pH 5.5) at different temperatures in the range of 30°C to 85°C. The thermostability of the purified β-xylosidase was monitored by preincubating 0.1 U of the enzyme in the absence of substrate at 60°C or 65°C. After being heated for the appropriate time interval, samples were cooled on ice and the residual enzyme activities were measured under standard assay conditions.

Effects of metal ions and other reagents

Various metal ions (1 mM), other reagents (1 mM) or
sugars (250 mM) were added to the standard enzymatic reaction mixtures containing 0.1 U of the purified enzyme in order to study their effects on β-xilosidase activity. Enzyme activities were then measured under standard assay conditions.

**Kinetic parameters**

The kinetic parameters of β-xilosidase for pNPX were determined by incubating the purified enzyme with different amounts of pNPX (0.2 to 2 mM) in sodium acetate buffer (pH 5.5) at 60°C. For each assay, ten different substrate concentrations were used in three independent experiments. The values of the Michaelis constant (K_m) and maximum velocity (V_max) were determined from Lineweaver-Burk plot.

**Hydrolysis studies**

The hydrolysis products from xylo-oligosaccharides (X_5, X_4 and X_3) by purified β-xilosidase were analyzed by thin-layer chromatography (TLC). The hydrolysis was carried out with 1 U of the purified β-xilosidase and 100 μg of xylo-oligosaccharides in 100 μl of 50 mM sodium acetate buffer (pH 5.5) at 60°C for 30 min. Equal amounts of the aliquots were removed periodically and the reaction was stopped by placing the mixture in boiling water for 10 min. A 1 μl portion of each sample was spotted onto a silica gel plate 60 F_254 (Merck, Darmstadt, Germany) and chro-

matographed in a solvent system containing n-butanol/acetate acid/water (2:1:1; v/v/v) at room temperature. The sugars on the plate were visualized by spraying the plate with 20% sulfuric acid and 0.1% orcinol in methanol, followed by heating at 110°C for 5 min.

**Results and Discussion**

**Purification of β-xilosidase**

An intracellular β-xilosidase was purified to homogeneity from the cell extract of *Paenibacillus* sp. DG-22 grown on 2x YT medium containing 1% (w/v) methyl β-D-xylopyranoside. The cell extract was fractionated by ion-exchange chromatography on Q-Sepharose. β-Xilosidase activity was eluted at 0.5 M NaCl using a 0-8 M NaCl gradient in 20 mM sodium phosphate buffer (pH 7.0). Active fractions were concentrated by ultrafiltration and dialyzed against 20 mM sodium phosphate buffer (pH 7.0) containing 0.6 M ammonium sulfate. In the second step the dia-

lyzed enzyme solution was loaded onto a Butyl Spherilose column pre-equilibrated with the same buffer. To obtain purified β-xilosidase, a further purification step with a gel-filtration chromatography on a Sephacryl S-300 column was performed. A summary of these three-step purification procedures is shown in Table 1. It was purified 64-fold with a 4.9% yield. The purified enzyme was shown to be homogeneous by the detection of a single protein band on SDS-PAGE (Fig. 1).

**Molecular weights of β-xilosidase**

The molecular mass of β-xilosidase in native state, based on the gel filtration chromatography, was 156 kDa (data not shown). SDS-PAGE analysis showed a single band with a molecular mass of about 80 kDa (Fig. 1, lane 4), suggesting that the enzyme consisted of two identical subunits. The molecular weight of β-xilosidase from *Paenibacillus* sp. DG-22 is within the range reported for other bacterial β-xilosidases. Bacterial β-xilosidases were

<table>
<thead>
<tr>
<th>Step</th>
<th>Total activity (U)</th>
<th>Total protein (mg)</th>
<th>Specific activity (U/mg)</th>
<th>Purification (folds)</th>
<th>Yield (%)</th>
</tr>
</thead>
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<tr>
<td>Sonic extract</td>
<td>372.4</td>
<td>116.4</td>
<td>3.2</td>
<td>100</td>
<td>100</td>
</tr>
<tr>
<td>Q-Sepharose</td>
<td>237.8</td>
<td>11.1</td>
<td>21.4</td>
<td>6.7</td>
<td>63.8</td>
</tr>
<tr>
<td>Butyl Spherilose</td>
<td>85.8</td>
<td>0.57</td>
<td>146.0</td>
<td>45.6</td>
<td>22.5</td>
</tr>
<tr>
<td>Sephacryl S-300</td>
<td>18.4</td>
<td>0.09</td>
<td>204.4</td>
<td>63.9</td>
<td>4.9</td>
</tr>
</tbody>
</table>

Fig. 1. SDS-PAGE analysis of the purified β-xilosidase from *Paenibacillus* sp. DG-22. Lane M, molecular mass marker; Lane 1, sonic extract; Lane 2, fraction after Q-Sepharose; lane 3, fraction after Butyl Spherilose; lane 4, fraction after Sephacril S-300 column.
Fig. 2. Effects of pH (A) and temperature (B) on the activity of purified β-xylanase from *Paenibacillus* sp. DG-22. The activities at the optimal pH and the optimal temperature were defined as 100%. Thermostability of purified enzyme at pH 5.5 in the absence of substrate (C). Residual activity was monitored at various times after incubation at 60°C (■) or 65°C (○). The initial activity was defined as 100%.

reported either as an homodimer of subunits having a molecular weight of 75-85 kDa [10,16,19] or an heterotrimer, comprising one subunit of 63 kDa and two of 85 kDa in the case of *Clostridium acetobutylicum* [12]. Molecular weights ranging from 39.5 kDa to 411 kDa have been reported for fungal β-xylanases, which behaved as a monomeric, dimeric, or tetrameric species [18].

**Effects of pH and temperature on β-xylanase activity**

The physicochemical properties of β-xylanase were determined with the purified enzyme. The enzyme had a rather sharp optimal pH and temperature. The pH optimum for hydrolysis of pNPX was 5.5, with 5% activity at pH 4.5 and 12% activity at pH 7.5 (Fig. 2A). The pH optimum is similar to those of other bacterial β-xylanases [10,16,19]. The optimum temperature for activity was 65°C with activity decreasing rapidly at higher temperature (Fig. 2B). The enzyme showed no activity at temperatures below 30°C and above 85°C. To examine the thermostability of the enzyme, β-xylanase was incubated at 60°C or 65°C for up to 60 min in the absence of substrate, and the residual activities were assayed (Fig. 2C). The purified enzyme was fairly stable at 60°C. It retained 80% of its initial activity up to 60 min of incubation. The enzymatic activity decreased significantly at 65°C. The half-life of the enzyme at 65°C was about 25 min.

**Substrate specificity**

The hydrolytic properties of β-xylanase toward various *p*-nitrophenyl glycosides were determined. The enzyme was most active against pNPX (Table 2). Lower or no activity was observed with the other substrates tested. Substrate

<table>
<thead>
<tr>
<th>Substrate (1 mM)</th>
<th>Relative activity (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>p-nitrophenyl-β-D-xlyopyranoside</td>
<td>100.0</td>
</tr>
<tr>
<td>p-nitrophenyl-α-D-xlyopyranoside</td>
<td>3.97</td>
</tr>
<tr>
<td>p-nitrophenyl-α-L-arabinofuranoside</td>
<td>0.43</td>
</tr>
<tr>
<td>p-nitrophenyl-α-L-arabinopyranoside</td>
<td>0.11</td>
</tr>
<tr>
<td>p-nitrophenyl-β-L-arabinopyranoside</td>
<td>0.11</td>
</tr>
<tr>
<td>p-nitrophenyl-β-D-fucopyranoside</td>
<td>0</td>
</tr>
<tr>
<td>p-nitrophenyl-α-D-glucopyranoside</td>
<td>0</td>
</tr>
<tr>
<td>p-nitrophenyl-β-D-glucopyranoside</td>
<td>0</td>
</tr>
<tr>
<td>p-nitrophenyl-β-D-mannopyranoside</td>
<td>0</td>
</tr>
</tbody>
</table>

ambiguity of β-xylanase has been reported for several β-xylanases. In *Trichoderma reesei* [17], *Butyrivibrio fibrisolvens* [21], *Thermomonospora fusca* [1] and *Thermoaerobacter ethanolicus* [19] the purified β-xylanases also showed α-arabinofuranosidase activity. The purified enzyme did not degrade oat spelt xylan, beechwood xylan and birchwood xylan (data not shown).

**Effects of metal ions and reagents on the enzyme activity**

The activity of β-xylanase was measured under standard assay conditions in the presence of metal ions and other reagents (Table 3). DTT increased the enzyme activity by 21%. On the other hand, AgCl completely inhibited the β-xylanase activity at a concentration of 1 mM. Addition of 1 mM FeCl₃, HgCl₂ or ZnCl₂ decreased the enzyme activity by 55%, 90% and 70%, respectively. The presence of EDTA did not affect the activity. These results suggest that metal ions are not necessary for the enzyme activity. The effects of arabinose, galactose, and xylose on the activity of β-xylanase were also investigated. Most of β-xylanases
Table 3. Effect of metal ions and some compounds on purified β-xylosidase

<table>
<thead>
<tr>
<th>Compounds</th>
<th>Concentration (mM)</th>
<th>Residual activity (%)</th>
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</thead>
<tbody>
<tr>
<td>Control</td>
<td></td>
<td>100.0</td>
</tr>
<tr>
<td>DTT</td>
<td>1</td>
<td>120.6</td>
</tr>
<tr>
<td>EDTA</td>
<td>1</td>
<td>103.3</td>
</tr>
<tr>
<td>β-mercaptoethanol</td>
<td>1</td>
<td>102.1</td>
</tr>
<tr>
<td>AgCl</td>
<td>1</td>
<td>0</td>
</tr>
<tr>
<td>CaCl₂</td>
<td>1</td>
<td>102.4</td>
</tr>
<tr>
<td>CoCl₂</td>
<td>1</td>
<td>96.8</td>
</tr>
<tr>
<td>CuCl₂</td>
<td>1</td>
<td>83.7</td>
</tr>
<tr>
<td>FeCl₂</td>
<td>1</td>
<td>45.1</td>
</tr>
<tr>
<td>HgCl₂</td>
<td>1</td>
<td>9.8</td>
</tr>
<tr>
<td>KCl</td>
<td>1</td>
<td>95.1</td>
</tr>
<tr>
<td>MgCl₂</td>
<td>1</td>
<td>97.6</td>
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<tr>
<td>MnCl₂</td>
<td>1</td>
<td>97.3</td>
</tr>
<tr>
<td>RbCl</td>
<td>1</td>
<td>97.1</td>
</tr>
<tr>
<td>ZnCl₂</td>
<td>1</td>
<td>30.1</td>
</tr>
<tr>
<td>Arabinose</td>
<td>250</td>
<td>74.9</td>
</tr>
<tr>
<td>Galactose</td>
<td>250</td>
<td>93.3</td>
</tr>
<tr>
<td>Xylose</td>
<td>250</td>
<td>55.9</td>
</tr>
</tbody>
</table>

Fig. 3. Thin layer chromatography of the hydrolysis products from xylo-oligosaccharides by purified β-xylosidase. The purified enzymes were incubated with xylobiose (X₂), xylotriose (X₃), and xylotetraose (X₄) in 50mM sodium acetate buffer (pH 5.5) at 60°C for 0 min (lane 0), 10 min (lane 1), 20 min (lane 2), and 30 min (lane 3), respectively. The standards (lane S) used were xylose (X₁), xylobiose (X₂), xylotriose (X₃), and xylotetraose (X₄).

studied were reported to be inhibited by xylose and some other monosaccharides such as arabinose and galactose [4,7]. As shown in Table 3, the enzyme activity decreased by 25%, 7% and 44% with arabinose, galactose and xylose, respectively at a 250 mM concentration.

Kinetic analysis of β-xylosidase
The kinetic parameters of β-xylosidase were investigated using pNPX as a substrate. The purified enzyme exhibited a typical Michaelis-Menten kinetics, with $K_m$ and $V_{max}$ values of 0.53±0.03 mM and 3.18±0.05 U/mg protein, respectively. The $K_m$ value was close to reported $K_m$ value of β-xylosidases from Bacillus sp. [10,20].

Hydrolysis of xylo-oligosaccharides
In order to confirm the mode of action of the β-xylosidase from Paenibacillus sp. DG-22, hydrolysis products from xylo-oligosaccharides ($X_2$, $X_3$, $X_4$) were analyzed by TLC (Fig. 3). The enzyme released xylose from all substrates suggesting that it is a true β-xylosidase.

Acknowledgement

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References

10. Lama, L., V. Calandrelli, A. Gambacorta and B. Nicolaus. 2004. Purification and characterization of thermostable xy-


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**초록**: *Paenibacillus* sp. DG-22로부터 β-xyllosidase의 정제 및 특성분석

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*Paenibacillus* sp. DG-22로부터 셰포네 호소인 β-xyllosidase가 이온교환, 소수성 상호작용, 칼럼과 크랙마토그래프에 의해 순수하게 정제되었다. 이 호소의 분자량은 젤과에 의해 156,000으로, SDS-PAGE에 의해 80,000으로 측정되었는데 이는 이 호소가 동일한 두 단위로 구성되어 있음을 나타낸다. 정제된 호소는 65℃와 pH 5.5에서 최대 활성을 나타내었다. 이 호소는 60℃에서부터 60분까지 초기 활성의 80%를 유지하였고 65℃에서 25분의 반감기를 가지고 있었다. 이 호소는 기질로서 pNPX에 매우 독이있었고 다른 p-nitrophenyl 글리코시드들과 자일란에는 활성을 나타내지 않았다. pNPX에 대한 Km과 Vmax는 각각 0.53 mM와 3.18 U/mg 단백질이었다. 이 β-xyllosidase는 Ca2+, Fe2+, Hg2+ 및 Zn2+에 의해 강하게 억제되었으며 DTT에 의해 약간 활성화되었다. 자일로바이오스, 자일로라모스 및 자일로레드라스로부터의 가수분해 산물은 자일로오스이었다.